

SYNTHESIS AND ANTIVIRAL ACTIVITY OF NOVEL ISODIDEOXY NUCLEOSIDES WITH EXOCYCLIC METHYLENE

Lak Shin Jeong* and Su Jeong Yoo

Laboratory of Medicinal Chemistry, College of Pharmacy, Ewha Womans University, Seoul 120-750, Korea

Received 11 December 1997; accepted 23 February 1998

Abstract: Novel isodideoxy nucleosides with exocyclic methylene were synthesized starting from L-xylose utilizing anomeric demethoxylation, Wittig reaction and Mitsunobu reaction as key steps and evaluated for antiviral activity. © 1998 Elsevier Science Ltd. All rights reserved.

Much attentions have been paid to unusual nucleosides since 1,3-dioxolanyl and 1,3-oxathiolanyl nucleosides were reported to be the promising anti-human immunodeficiency virus (HIV) and anti-hepatitis B virus (HBV) agents.¹⁻⁴ Among these compounds, (-)-L-β-1,3-oxathiolanyl cytosine (**1a**, 3TC, Lamivudine) is being clinically used as anti-AIDS drugs and will be soon approved by Food and Drug Administration (FDA) for the treatment of HBV infected individuals as well.² (-)-L-β-1,3-Dioxolanyl cytosine (**1b**) also exhibited extremely potent anti-HIV and anti-HBV activities, but its cytotoxicity hindered it from being further developed as antiviral agent.⁴

Recently, carbocyclic nucleoside 2 with an exocyclic methylene group in place of oxygen atom of the furanose ring was reported to show antiviral activity, especially anti-HBV activity.⁵

Based on these findings, we wanted to synthesize the novel compounds 3, which replace C-OH of the 3-position in compound 2 with bioisosteric oxygen atom, 4 that would combine the properties of L-dioxolanyl nucleosides and exocyclic methylene substituted nucleosides. Here, we report synthesis and antiviral activity of novel isodideoxynucleosides with an exocyclic methylene substituent starting from L-xylose.

Synthesis of the key intermediate 11 for the synthesis of isodideoxy nucleosides with exocyclic methylene is illustrated in Scheme 1. L-Xylose was converted to compound 5 according to the known method.⁶ Isopropylidene group of 5 was removed using 70% acetic acid (50 °C, 1 h) to give the diol 6. The selective protection of primary hydroxyl group was achieved by dibutyltin oxide method.⁷ Treatment of compound 6 with dibutyltin oxide in refluxing toluene for 5 h followed by addition of benzyl bromide and *n*-tetrabutylammonium bromide (100 °C, 15 h) afforded dibenzyl derivative 7 in almost quantitative yield. Next step was the removal of anomeric methoxy group. We first benzoylated secondary hydroxyl group in compound 7 and then treated with triethylsilane and TMSOTf in CH₂Cl₂ to give the demethoxylated compound,⁸ but this method needed extra debenzoylation step to prepare the desired compound 8 (61% from 7). To eliminate extra benzoylation and debenzoylation steps, we used *in situ* silylation method.⁹ Refluxing 7 with hexamethyldisilazane (HMDS) followed by treatment with triethylsilane and TMSOTf in CH₂Cl₂ at room temperature for 2 h gave the demethoxylated compound 7 in 80% yield directly. The secondary hydroxyl group of 8 was oxidized with DMSO and acetic anhydride (rt, 18 h) to the ketone (80%). Wittig

reaction of the ketone intermediate with BuLi and Ph₃PCH₃Br at 0 °C produced the olefin **9** in 30-40% yield, while use of t-amyl alcohol (rt, 0.5 h) and NaH¹⁰ instead of BuLi afforded the same product **9** in 92% yield. Debenzylation of **9** with BCl₃ at -78 °C for 0.5 h gave the diol **9**, whose primary hydroxyl group was selectively protected with *t*-butyldiphenylsilyl (TBDPS) group to yield **11**,¹¹ which acts as the key intermediate for the desired isodideoxy purine and pyrimidine nucleosides.

Synthesis of the purine nucleosides having adenine, hypoxanthine and guanine was accomplished using the Mitsunobu reaction (Scheme 2). ¹² Treatment of 11 with DEAD and PPh₃ in THF at 0 °C for 1 h produced the N-9 derivative 12 in 80% yield without the formation of N-7 isomer. The N-9 isomer of the coupling was confirmed by UV spectral data [λ_{max} (MeOH) 264 nm]. ^{12a} Desilylation (Bu₄NF, THF, 0 °C, 0.5 h) of 12 followed by amination with methanolic ammonia (100 °C, 15 h) afforded the adenine derivative 3a. ¹³ Compound 14 was converted to hypoxanthine derivative 3b by refluxing with 1 N NaOH. For the synthesis of guanine analogue, compound 11 was reacted with 2-amino-6-chloropurine using the same Mitsunobu conditions to give 13 (50%) with concomminant formation of the N-7 substituted product (5%). Compound 13 was successively treated with Bu₄NF and 1 N NaOH to afford the guanine analogue 3c. The regioisomers were also confirmed by comparison of the UV data of N-9 (252 nm) and N-7 (248 nm) guanine analogues. ¹⁴

Scheme 3

TBDPSO 11
$$\frac{O}{DEAD, PPh_3}$$
THF 63%

TBDPS = $-\frac{Ph}{Si}$
TBDPS = $-\frac{Si}{C}(CH_3)_3$
Ph 75% 1) NaOMe, MeOH 2) Bu₄NF, THF

NH₂
1) Ac₂O, pyridine 2) 1,2,4-triazole, POCl₃, NEt₃, CH₃CN
3) NH₄OH : dioxane (1:4) 4) NH₃, MeOH 100%

4b 4a

Isodideoxy pyrimidine nucleosides (4a and 4b) were also synthesized utilizing the Mitsunobu reaction (Scheme 3). The key intermediate 11 was condensed with N³-benzoyluracil under the standard Mitsunobu conditions¹² to give the desired N-alkylated product 16 (63%) with concomminant formation of O-alkylated compound (10%). The regioisomers were easily confirmed by comparison of the UV literature data.¹²a Protecting groups of 16 were removed by treating successively with Bu₄NF and NaOMe successively to yield uracil derivative 4a.¹⁵ Finally, the stereochemistry of the C2-position in compound 4a was decided by NOSEY experiment, indicating the Mitsunobu reaction of the allylic alcohol 11 was proceeded in pure S_N2 type reaction, not in S_N¹ or S_N² type reaction.¹⁶ Uracil analogue 4a was converted to the cytosine derivative 4b according to the conventional method.

The antiviral assays against human immunodeficiency virus 1 (HIV-1), herpes simplex virus-1,2 (HSV-1,2) and human cytomegalovirus (HCMV) were performed and the results are shown in Table 1. As shown in Table 1, all synthesized compounds did show neither anti-HIV activity nor cytotoxicity. Any compounds did not show antiviral activity against HSV-1,2 except hypoxanthine derivative 3b which exhibited very weak anti-HSV-1 activity. However, the uracil analogue 4a was found to show significant anti-HCMV activity and the adenine derivative 3a also exhibited weak anti-HCMV activity.

In summary, we completed the synthesis of bioisosteric compounds (3a, 3b, 3c, 4a, and 4b) of potent antiviral agent 2, starting from L-xylose utilizing demethoxylation, Wittig reaction and Mitsunobu reaction as key steps. The hypoxanthine derivative 3b exhibited weak anti-HSV-1 activity and the uracil derivative 4a exhibited significant anti-HCMV activity.

Acknowledgment

Antiviral testing by Dr. Chong-Kyo Lee (Korea Research Institute of Chemical Technology) is greatly appreciated. Authors also thanks KOSEF for the financial support of this research.

Activity	HIV-1	HSV-1	HSV-2	HCMV	cytotoxicity
Compounds	EC ₅₀ (µg/ml)	EC ₅₀ (µg/ml)	EC ₅₀ (μg/ml)	EC ₅₀ (µg/ml)	IC ₅₀ (μg/ml)
3a	> 100	> 100	> 100	33.3	> 100
3b	> 100	35	> 100	> 100	> 100
3c	> 100	> 100	> 100	> 100	> 100
4a	> 100	> 100	> 100	10.6	> 100
4b	> 100	> 100	> 100	> 100	> 100
AZT	0.00132	ND	ND	ND	1.0
Acyclovir	ND	1.0539	5.1165	ND	250
Ganciclovir	ND	ND	ND	0.74	> 10

Table 1. The antiviral activities of the synthesized compounds.

ND: Not Determined

References and Notes

- (a) Belleau, B.; Dixit, D.; Nguyen-Ga, N.; Kraus, J. L. 5th International Conference on AIDS, Montreal, Canada, June 4-9, 1990, paper no. T.C.O.I. (b) Soudeyns, H.; Yao, Q.; Gao, Q.; Belleau, B.; Kraus, J.-L.; Nguyen-Ga, N.; Spira, B.; Wainberg, M. A. Antimicrob. Agents Chemother. 1991, 35, 1386.
- 2. Norbeck, D. W.; Spanton, S.; Broder, S.; Mitsuya, H. Tetrahedron Lett. 1989, 33, 6263.
- 3. (a) Beach, J. W.; Jeong, L. S.; Alves, A. J.; Pohl, D.; Kim, H. O.; Chang, C.-N.; Doong, S.-L.; Schinazi, R. F.; Cheng, Y.-C.; Chu, C. K. J. Org. Chem. 1992, 57, 2217. (b) Jeong, L.S.; Schinazi, R. F.; Beach, J. W.; Kim, H. O.; Nampalli, S.; Shanmuganathan, K.; Alves, A. J.; McMillan, A.; Chu, C. K. J. Med. Chem. 1993, 36, 181 and references cited therein.
- 4. Kim, H. O.; Shanmuganathan, K.; Alves, A. J.; Jeong, L. S.; Beach, J. W.; Schinazi, R. F.; Chang, C.-N.; Cheng, Y.-C.; Chu, C. K. *Tetrahedron Lett.* 1992, 33, 6899 and references cited therein.
- 5. Bisacchi, G. S.; Chao, S. T.; Bachard, C.; Daris, J. P.; Innaimo, S.; Jacobs, G. A.; Kocy, O.; Lapointe, P.; Martel, A.; Merchant, Z.; Slusarchyk, W. A.; Sundeen, J. E.; Young, M. G.; Colonno, R.; Zahler, R. Bioorg. & Med. Chem. Lett. 1997, 7, 127.

- 6. El Khadem, H. S. Nucleic Acid Chemistry, Part 1, Townsend, L. B.; Tipson, R. S. (Editors), 1978, pp169-182.
- 7. David, S.; Hanessian, S. Tetrahedron 1985, 41, 643.
- 8. Jeffery, A.; Nair, V. Tetrahedron Lett. 1995, 36, 3627.
- 9. Bennek, J. A.; Gray, G. R. J. Org. Chem. 1987, 52, 892.
- 10. Yoshimura, Y.; Kitano, K.; Satoh, H.; Watanabe, M.; Miura, S.; Sakata, S.; Sasaki, T.; Matsuda, A. J. Org. Chem. 1996, 61, 822.
- 11. Compound 11: 1 H NMR (CDCl₃, 250 MHz) δ 7.70–7.25 (m, 10 H, Ph x 2), 5.36 (t, 1 H, J = 2.0 Hz, vinyl), 5.11 (t, 1 H, J = 1.9 Hz, vinyl), 4.70-4.60 (m, 2 H, 2-H and 4-H), 4.13 (dd, 1 H, J = 5.4, 9.3 Hz, 1-H_a), 3.80-3.40 (m, 3 H, 1-H_b and 5-H), 1.04 (s, 9 H, *t*-butyl).

 Anal. Calcd for $C_{22}H_{28}O_{3}Si: C$, 71.74; H, 7.61. Found: C, 71.75; H, 7.66.
- 12. (a) Bannal, C.; Chavis, C.; Lucas, M. J. Chem. Soc. Perkin Trans 1 1994, 1401. (b) Jenny, T. F.; Horlacher, J.; Previsani, N.; Benner, S. A. Helv. Chim. Acta 1992, 75, 1944.
- 13. Compound 3a: UV (MeOH) λ_{max} 259 nm; ¹H NMR (CD₃OD, 300 MHz) δ 8.31 (s, 1 H, H-8), 8.22 (s, 1 H, H-2), 5.66 (m, 1 H, 2'-H), 5.36 (m, 2 H, vinyl), 4.54 (m, 1 H, 4'-H), 4.22 (pseudo t, 2 H, J = 3.9, 9.5 Hz, 1'-H), 3.95 (dd, 1 H, J = 3.2, 12.2 Hz, 5'-H_a), 3.87 (dd, 1 H, J = 4.1, 12.2 Hz, 5'-H_b). Anal. Calcd for C₁₁H₁₃O₂N₅: C, 53.44; H, 5.30; N, 28.34. Found: C, 53.75; H, 5.60; N, 28.02.
- 14. Albert, A. in *Synthetic Procedures in Nucleic Acid Chemistry*, eds. Zorbach, W. W. And Tipson, R. S. **1973**, vol.2, pp. 47-123.
- 15. Compound **4a**: UV (MeOH) λ_{max} 265 nm; ${}^{1}H$ NMR (DMSO-d₆, 250 MHz) δ 11.42 (br s, 1 H, NH), 7.73 (d, 1 H, J = 8.1 Hz, H-6), 5.67 (d, 1 H, J = 8.1 Hz, H-5), 5.54 (m, 1 H, 2'-H), 5.38 (m, 2 H, vinyl), 5.12 (t, 1 H, J = 5.5 Hz, OH, exchangeable with D₂O), 4.42 (m, 1 H, 4'-H), 4.08 (dd, 1 H, J = 6.8, 9.8 Hz, 1'-H_a), 3.97 (dd, 1 H, J = 4.4, 9.8 Hz, 1'-H_b), 3.82 (m, 2 H, 5'-H). Anal. Calcd for C₁₀H₁₂O₄N₂: C, 53.57; H, 5.39; N, 12.49. Found: C, 53.65; H, 5.66; N, 12.29.
- 16. Shull, B. K.; Sakai, T.; Nichols, J. B.; Koreeda, M. J. Org. Chem. 1997, 62, 8294.